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Hypocitraturia: its importance as a factor in the development of urolithiasis

Hipocytraturia – znaczenie w rozwoju kamicy układu moczowego

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Abstract

The citrate anion is an endogenous inhibitor of calcium deposits in the urinary tract. Moreover, by urine alkalinisation, citrates enhance uric acid and cystine solubility. Hypocitraturia in adults is defined as excretion of citrates with urine below 320 mg/24 h. The most common definitions in children involve citrate excretion in 24 hour urine collection at a level <365 mg/1.73 m²/24 h in boys and <310 mg/1.73 m²/24 h in girls. Moreover, the urine citrate/creatinine ratio is also considered: for children aged 0–5 years <0.42 mg/mg and for children older than 5 years of age <0.25 mg/mg. Hypocitraturia is a common metabolic disturbance; such a diagnosis is made in approximately 20–60% of adults and 10–68% of children with urolithiasis. Various factors may affect citrate excretion with urine, the most significant of which is the pH value (of blood, urine or intracellular fluid). Most patients are diagnosed with so-called idiopathic hypocitraturia, although it also accompanies acidosis, including distal tubular acidosis, hypokalaemia, high-protein diet and treatment with certain medications (carbonic anhydrase inhibitors, angiotensin-converting enzyme inhibitors, thiazide diuretics). Patients with hypocitraturia should first and of foremost have dietary modifications introduced. This involves an increase in the dietary amount of fluids and citrus fruit as well as salt and protein restriction. Patients with recurring urolithiasis may also be recommended citrate products, the most common of which is potassium citrate. Citrus juices may be an alternative, but their effect is not as potent and unambiguous as in the case of commercially available citrate products.

Keywords: hypocitraturia, citrates, urolithiasis, children

Streszczenie

Anion cytrynianowy to endogenny inhibitor powstawania złogów wapniowych w drogach moczowych. Ponadto przez alkalizację moczu cytryniany zwiększają rozpuszczalność kwasu moczowego i cystyny. Hipocytraturia definiowana jest u dorosłych jako wydalanie cytrynianów z moczem poniżej 320 mg/24 h. Najczęściej stosowane definicje hipocytraturii u dzieci to wydalanie cytrynianów w dobowej zbiórce moczu <365 mg/1,73 m²/24 h u chłopców i <310 mg/1,73 m²/24 h u dziewczynek; używa się też wskaźnika cytrynianowo-kreatyninowego z porcji moczu: dla dzieci w wieku 0–5 lat <0,42 mg/mg, a powyżej 5. roku życia: <0,25 mg/mg. Hipocytraturia to częsta nieprawidłowość metaboliczna – stwierdza się ją u około 20–60% dorosłych i 10–68% dzieci z kamicy. Na wydalanie cytrynianów z moczem wpływ ma wiele czynników, z których najważniejszym jest wartość pH (krwi, moczu oraz wewnątrzkomórkowe). Większość pacjentów ma tzw. idiopatyczną hipocytraturię, choć występuje ona także w stanach kwasicy, w tym kwasicy kanalikowej dystalnej, hipokaliemii, diecie bogatobiałkowej oraz przy stosowaniu niektórych leków (inhibitory anhidrazy węglanowej, inhibitory konwertazy angiotensyny, diuretyki tiazydowe). U pacjentów z hipocytraturią w pierwszej kolejności należy podjąć modyfikacje dietetyczne, polegające na zwiększeniu ilości przyjmowanych płynów i owoców cytrusowych oraz ograniczeniu spożycia soli i białka. U pacjentów z nawrotową kamicy zastosowanie znajdują preparaty cytrynianowe, z których najczęściej stosuje się cytrynian potasu. Alternatywnie można stosować soki cytrusowe, jednak ich korzystne działanie nie jest tak silne i jednoznaczne jak w przypadku komercyjnie dostępnych preparatów cytrynianów.

Słowa kluczowe: hipocytraturia, cytryniany, kamica moczowa, dzieci

UROLITHIASIS – HYPOCITRATURIA

According to epidemiological studies, a decreased level of crystallisation inhibitors is one of the most significant risk factors of urolithiasis⁽¹⁾. Citrates are substances deemed the most important urinary calculus inhibitors. Their reduced urinary excretion (hypocitraturia) may be an isolated defect in patients with urolithiasis, but may also accompany hypercalciuria or hyperoxaluria. Citrates are inhibitors of calcium oxalate and calcium phosphate crystallisation. Hypocitraturia is identified in approximately 20–60% of adults⁽²⁾ and 10–68% of children with urolithiasis^(3–7). In the analysis of 113 patients with nephrolithiasis, hypocitraturia was identified in 68% of children, including 81% with a genetically or systemically determined nephrolithiasis (primary hyperoxaluria, cystinuria, Dent disease, 2,8-dihydroxyadenine urolithiasis, tubular acidosis, Lesch–Nyhan disease, Crohn's disease, ketogenic diet) and 65% with idiopathic nephrolithiasis⁽⁵⁾. Moreover, hypocitraturia was noted in 31.7% (19/60) of children with asymptomatic microscopic haematuria⁽⁶⁾. In 63 children with nephrolithiasis, hypocitraturia was the most common (58.1%) metabolic abnormality, even more common than hypercalciuria. The authors put forward a hypothesis that hypocitraturia is becoming the most common risk factor of urolithiasis in the paediatric population due to unfavourable dietary changes⁽⁷⁾. It is the primary risk factor of urolithiasis in preterm infants⁽⁸⁾ and in children with very low birth weight⁽⁹⁾ as well as in some parts of the world (for example in Turkey)⁽¹⁰⁾.

Citrate excretion is higher in children than in adults. Hypocitraturia in adults is identified with citrate excretion below 320 mg/24 h (=1.67 mmol/24 h)⁽¹¹⁾. The diagnostic criteria of hypocitraturia in children according to Hoppe and Kemper are⁽⁸⁾:

1. amount of citrate in 24 h urine collection: boys: <365 mg/1.73 m²/24 h; girls: <310 mg/1.73 m²/24 h;
2. citrate/creatinine ratio in a urine sample:
0–5 years of age: <0.42 mg/mg (<0.25 mol/mol);
>5 years of age: <0.25 mg/mg (<0.15 mol/mol).

It must be underlined that the reference levels are arbitrary, and the risk of urolithiasis grows continuously with a decreased urine citrate level⁽¹²⁾.

Kirejczyk et al. evaluated urinary citrate excretion in 2,334 healthy boys and girls aged 2–18 years. The authors showed that citrate excretion, expressed in absolute values, increases with age and is higher in boys than in girls. Based on these results, they have proposed the following diagnostic criteria for hypocitraturia (defined as citrate excretion <5th percentile in the studied population): boys: 180 mg/1.73 m²/24 h; girls: 250 mg/1.73 m²/24 h⁽¹³⁾.

Hypocitraturia may be the only risk factor of urolithiasis, but often coexists with other potential causes, usually with hypermagnesuria, but also with hypercalciuria and hyperoxaluria or cystinuria. It therefore seems that the estimation of reciprocal relationships of calculus

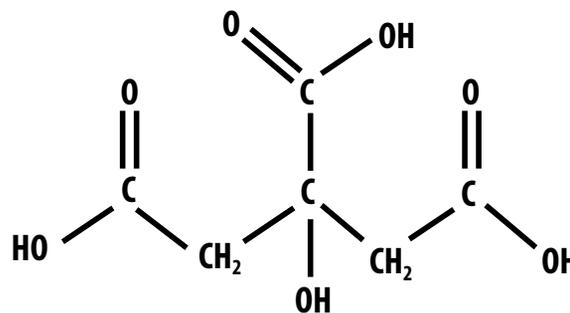


Fig. 1. Chemical structure of citric acid^(62, own modification)

components and calculus inhibitors (e.g. citrates) may be useful. DeFoor et al. demonstrated that the calcium-to-citrate ratio may occur adequate for this purpose. The ratio amounted to 0.33 in healthy children, 0.41 in children with one episode of urolithiasis and 0.64 in children with recurring urolithiasis⁽¹⁴⁾. Turudic et al., however, found the oxalate/(citrate × glycosaminoglycans) ratio to be an independent risk factor of calculus formation in children⁽¹⁵⁾.

CITRATE METABOLISM

Citric acid is tricarboxylic acid (Fig. 1) synthesised in the mitochondria from oxaloacetate and acetyl-CoA with the participation of citrate synthase (Fig. 2). This is the central stage of tricarboxylic acid cycle (also called Krebs cycle or citric acid cycle), which is a common terminal pathway for oxidation of carbohydrates, proteins and lipids⁽¹⁶⁾. The serum citrate concentration persists at a relatively

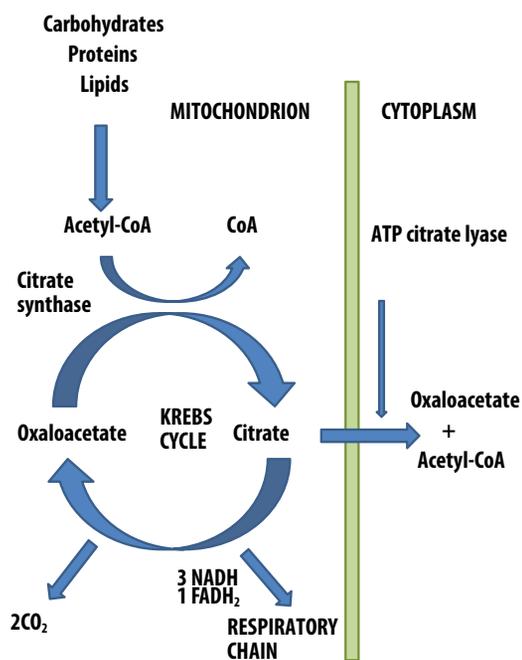


Fig. 2. Metabolism of citric acid^(62, own modification)

stable level: from 0.1 to 0.15 mmol/L⁽¹⁷⁾. Approximately 90% of citrates in the body are located in bones, which are their main reservoir, with osteoblasts being the basic site of citrate production. Citrates are released to circulation during bone tissue resorption⁽¹⁷⁾. Most circulating citrates assume the form of complexes with divalent cations, such as calcium and magnesium. It is believed that the most important hormones responsible for maintaining an appropriate blood citrate concentration are parathyroid hormone and calcitonin owing to their impact on bone metabolism⁽¹⁷⁾.

Citrates are filtrated in the renal glomeruli and enter the primary urine. Citrate reabsorption occurs mainly at the level of the proximal tubule⁽¹⁸⁾. In humans, 65–90% of filtered citrates are reabsorbed⁽¹⁹⁾. That is why their urinary excretion mainly depends on the degree of proximal tubule reabsorption⁽¹⁸⁾. The most important protein responsible for citrate reabsorption is sodium-dependent dicarboxylate transporter 1 (NaDC-1), whose expression is noted in the proximal tubule, small and large bowels, liver and brain⁽¹⁹⁾. NaDC-1 exerts activity towards dinegative citrates but is inhibited by trinegative citrates. Energy needed for citrate reabsorption is provided by the sodium–potassium pump, located on the surface of the basolateral cells of the proximal tubule. Most citrates that cross the apical membrane enter the mitochondria and are used in the tricarboxylic acid cycle⁽¹⁸⁾. Citrates that remain in the cytoplasm are metabolised by ATP citrate lyase. It catalyses citrate conversion to oxaloacetate and acetyl-CoA using coenzyme A and ATP. Acetyl-CoA is subsequently utilised, for

instance in the process of fatty acid and cholesterol synthesis, while oxaloacetate is a substrate for gluconeogenesis⁽²⁰⁾. Non-metabolised citrates return to circulation via the renal vein⁽¹⁷⁾.

Renal citrate reabsorption is regulated by various factors, the most important being pH (of urine, plasma, renal interstitium and intracellular fluid). Acidosis reduces renal citrate excretion, while alkalosis enhances it⁽²¹⁾. One of the mechanisms responsible for the impact of pH is the degree of citrate ionisation. Dinegative citrates transported by NaDC-1 are present in higher concentrations in acidic pH compared to alkaline pH. Chronic acidoses not only enhance the activity of NaDC-1, but also increase its expression. Moreover, these conditions intensify intratubular citrate metabolism⁽¹⁹⁾.

Bowel citrate absorption is also mediated by NaDC-1. Citrate bioavailability following oral administration is 96–98%⁽²¹⁾.

MECHANISM OF ANTI-UROLITHIASIS EFFECTS OF CITRATES

The mechanisms of anti-urolithiasis effect of citrates and their renal transport are illustrated in Fig. 3. Citrates are bound with calcium in the renal tubules, thereby forming a non-dissociating but soluble complex. This decreases the pool of free calcium, available for binding with oxalates and phosphates. Citrates also seem to inhibit crystal enlargement (aggregation), which is a process whereby single calcium oxalate crystals combine into a calculus⁽²²⁾. Additionally, citrates bind with the surface of a calculus. This prevents calcium oxalate adhesion to epithelial cells⁽²³⁾. Furthermore, the mechanism of action of citrates consists in entering into a positive interaction with another crystallisation inhibitor, namely Tamm–Horsfall protein (uromodulin)⁽²⁴⁾. It has also been demonstrated that they reduce the urine level of osteopontin, which is a common component of a calculus. Moreover, they improve uric acid and cystine solubility by increasing urine pH⁽²⁵⁾.

CLINICAL CONDITIONS LEADING TO HYPOCITRATURIA

Clinical conditions leading to hypocitraturia are presented in Tab. 1.

Idiopathic hypocitraturia, dietary factors and genetic factors

Although factors of idiopathic hypocitraturia are unknown, the proposed aetiology is associated with environmental factors (mostly with high-protein diet)⁽²⁶⁾ and genetic predisposition. High-protein diet (especially with protein from meat, poultry and eggs) acidifies the system and increases citrate reabsorption at the level of the proximal tubule. It has been shown that low-carbohydrate acidifying diets

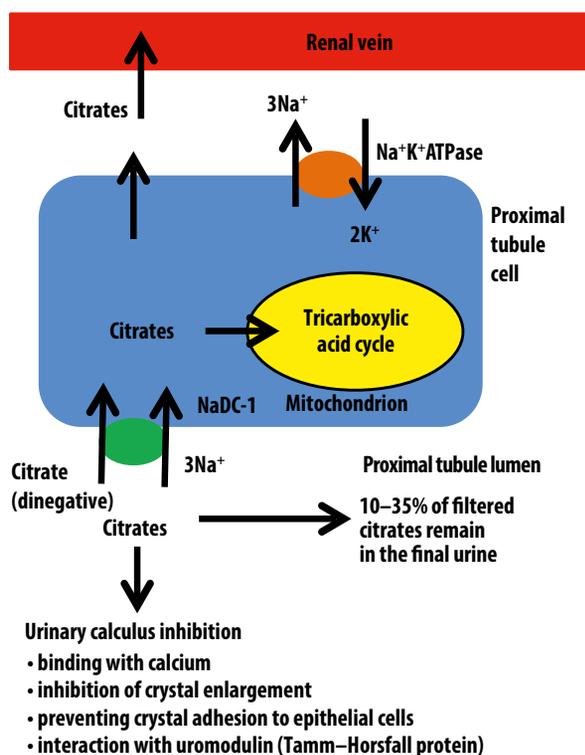


Fig. 3. Citrate renal transport and mechanisms of their anti-urolithiasis action^(61, own modification)

Metabolic acidosis	Distal tubular acidosis Metabolic acidoses of other aetiology (e.g. inborn errors of metabolism) Chronic diarrhoeas Malabsorption syndromes Intensive physical effort
Systemic/renal ionic disorders	Hypokalaemia Hypokaliuria Hypomagnesuria
Hormonal disorders	Hyperparathyroidism Hyperaldosteronism
Dietary factors	High-protein diet High-salt diet Ketogenic diet Diet poor in vegetables/fruit Fasting
Drugs	Carbonic anhydrase inhibitors Angiotensin-converting enzyme inhibitors Thiazide diuretics
Other medical conditions	Glycogen storage disease type I
Genetic factors	Vitamin D receptor (VDR) gene polymorphisms Sodium-citrate transporter (NaDC-1) gene polymorphisms ZNF365 (zinc finger) gene polymorphisms

Tab. 1. Clinical conditions leading to hypocitraturia^(61, own modification)

(such as Atkins diet, ketogenic diet or Kwaśniewski diet) also reduce urinary citrate excretion⁽²⁷⁾.

In addition, fruit-poor diet results in body acidification, a low intake of potassium and low urinary potassium excretion, which correlates with urinary citrate excretion^(26,28). Furthermore, it has been demonstrated that hypocitraturia is linked with a high-sodium diet⁽²⁹⁾ and fasting⁽³⁰⁾ (in both cases, probably in the mechanism of mild metabolic acidosis).

A relationship has been noted also between vitamin D receptor (VDR) gene polymorphisms and urinary citrate excretion⁽³¹⁾. The active form of vitamin D, i.e. 1.25(OH)₂D, modulates NaDC-1 activity by binding with the VDR receptor⁽³²⁾.

Moreover, it has been observed that urinary citrate excretion is correlated with polymorphisms of the sodium-citrate cotransporter (hNaDC-1) gene and zinc finger protein 365 (ZNF365) gene that encodes talanin⁽³⁴⁾.

Electrolyte and tubular disorders

Metabolic acidoses

Children with chronic metabolic acidosis are at a higher risk of urolithiasis. Children with acidosis caused by increased citrate reabsorption in the proximal tubule excrete lower amounts of citrates with urine as these citrates are used for the regeneration of the alkaline pool in the body. In the mechanism of chronic metabolic acidosis, hypocitraturia, which entails a higher risk of urolithiasis, also occurs in chronic diarrhoeas (e.g. in cystic fibrosis)⁽³⁵⁾.

Moreover, hypocitraturia is common in (type 1) distal tubular acidosis. Exacerbation of hypocitraturia is higher in complete tubular acidosis, but may also occur

in incomplete distal renal tubular acidosis (dRTA) with the absence of overt metabolic acidosis⁽³⁵⁾. It must be noted that other types of tubular acidosis, such as the proximal form (type 2) and hypokalaemic form (type 4), are not associated with hypocitraturia. Nevertheless, this disorder has been identified in children with secondary tubular acidosis, e.g. after treatment with ifosfamide, or in substance addicts “sniffing glue” (chronic toluene poisoning).

Patients with malabsorption syndromes are also diagnosed with hypocitraturia, which is secondary not to impaired citrate absorption, but rather to intestinal carbohydrate loss⁽³⁶⁾.

Hypokalaemia

Hypokalaemia enhances NaDC-1 activity by leading to intratubular and intracellular acidosis (in the mechanism of transmembrane K⁺ and H⁺ exchange)⁽²⁸⁾.

Hypercalciuria and hypermagnesuria

Increased urinary excretion of calcium and magnesium also enhances urinary citrate excretion⁽³⁷⁾. Kovacevic et al. showed that potassium and magnesium excretion is positively correlated with citrate excretion⁽⁷⁾. Administration of magnesium increases urinary citrate excretion⁽³⁸⁾. Calcium and magnesium bind with citrates in urine, thereby decreasing their pool available for NaDC-1.

Hormonal factors

Parathyroid hormone reduces urinary citrate excretion⁽¹⁷⁾. Hypocitraturia is detected in approximately 20–40% of patients with hyperparathyroidism⁽³⁹⁾. A precise mechanism of action of parathyroid hormone remains unclear. There are no data on its effect on increasing NaDC-1 activity. However, it has been shown that it may inhibit citrate metabolism in tubular cells in the kidneys⁽⁴⁰⁾.

Animal studies indicate that renal citrate excretion is increased by calcitonin. Its administration to rats caused a five-fold increase in citraturia⁽⁴¹⁾. The mechanism of action of this phenomenon and its relevance in humans remain unclear.

Drugs

Hypocitraturia is identified in patients treated with carbonic anhydrase inhibitors. These drugs induce systemic acidosis by inhibiting carbohydrate reabsorption at the level of the proximal tubule. This, in turn, results in increased citrate reabsorption from urine. Drugs from this group include: acetazolamide, topiramate and zonisamide⁽³⁶⁾. Hypocitraturia has been noted in 93% of children treated with topiramate⁽⁴²⁾.

The disorder may also appear secondary to treatment with angiotensin-converting enzyme inhibitors. It is believed that the mechanism of action of these drugs consists in increasing the activity of ATP citrate lyase in cells of the proximal tubule⁽⁴³⁾.

Thiazide diuretics reduce urinary citrate excretion probably in the mechanism of hypokalaemia induction^(26,44). Lithium, on the other hand, increases urinary citrate levels by suppressing their reabsorption in the proximal tubule⁽⁴⁵⁾. Vitamin D also affects the activity of NaDC-1⁽³²⁾ and citrate metabolism in tubular cells⁽¹⁹⁾, which results in increased urinary citrate excretion.

Other clinical conditions

The concentration of citrates in the primary urine decreases with a decreasing glomerular filtration in chronic kidney disease. However, an increase in the filtration fraction maintains the stable citrate concentration in the final urine until the stage of severe renal failure⁽⁴⁶⁾. Hyperaldosteronism increases the sodium pool in the body and causes hypokalaemia, thereby inducing hypocitraturia and hypercalciuria⁽⁴⁷⁾. Hypocitraturia (as hypercalciuria and hyperuricosuria) has been identified in patients with glycogen storage disease type I⁽⁴⁸⁾. Despite increased risk of nephrolithiasis in obesity, the relationship between body mass index and citraturia is not clear^(26,49). Physical exercise does reduce urinary citrate excretion, but water reabsorption maintains their relatively stable levels⁽⁵⁰⁾.

THERAPEUTIC MANAGEMENT IN HYPOCITRATURIA

The risk of urolithiasis is associated with dietary modifications. Increased intake of fruit and vegetables has a positive impact on urinary citrate excretion⁽⁵¹⁾. The dual mechanism of action in this case consists in the presence of citrates in these products and in increasing the potassium and magnesium pool. It must also be underlined that ingested citrates are mainly metabolised in the liver to bicarbonates, and they inhibit urinary citrate reabsorption by increasing pH⁽²⁵⁾.

Oral citrate supplementation reduces the risk of urolithiasis in patients with hypocitraturia⁽⁵²⁾. Potassium citrate is the most common supplement. These products increase the alkaline pool and raise urinary pH, thereby reducing the risk of uric acid urolithiasis and cystine urolithiasis. The efficacy of potassium citrate supplementation has also been shown in patients with infection nephrolithiasis⁽⁵³⁾.

The effects of citrate supplementation have been evaluated in prospective studies in adults with recurring urolithiasis^(54,55). Doses ranged from 60 to 90 mmol/24 h. The efficacy of potassium citrate (administered at a dose of 1 mmol/kg/24 h) for prevention of recurring episodes of urolithiasis has been demonstrated also in children with hypocitraturia⁽⁵⁶⁾. In 82 children with hypocitraturia, recurring episodes of urolithiasis were observed in only 1 of 71 children (1.4%) treated with potassium citrate and in all 11 children who received no supplementation⁽⁵⁶⁾. Undesirable effects of potassium citrates are mild and usually involve the gastrointestinal tract (bloating,

nausea, diarrhoea)⁽⁵⁷⁾. They are more common with liquid preparations than tablets. Nevertheless, the rate of citrate absorption from tablets is lower compared to liquid preparations.

The citrate dose depends on patient's body mass and the degree of hypocitraturia and acidosis. Typical doses for adults range from 40 to 60 mmol/24 h. In children, the recommended doses are 0.1–0.2 g/kg/24 h (0.3–0.6 mmol/kg/24 h). They may be significantly higher in patients with distal tubular acidosis, where the minimum dose of potassium citrate is usually 0.2–0.3 g/kg/24 h.

Currently, there are two products with potassium citrate available in Poland. Tablets of 680 mg (1 tablet containing 2.2 mmol) (Litocid) are registered for treatment of adults with calcium urolithiasis and low urinary citrate excretion. A recommended starting dose is 3 × 2 tablets (approximately 50 mg/kg/24 h), which is subsequently modified depending on urine pH. Granules (Citrolyt) contain 46.4 g of potassium citrate monohydrate, 39.1 g of sodium citrate dihydrate and 14.5 g of citric acid monohydrate in 100 g (i.e. 3.16 mmol of citrates in 1 g). The product is registered for treatment of uric acid and urate nephrolithiasis in adults and children over 12 years of age. The initial dosing pattern is: 2.5 g in the morning at 7–8 a.m., next 2.5 g at 2–3 p.m. and 5.0 g at 10 p.m. Further treatment is conducted on the basis of urine pH values.

Patients administered citrates require monitoring of electrolyte levels, urine pH as well as mineral and citrate excretion levels. The urine pH after administration of citrates may increase to the level that is a risk of calcium–phosphate urolithiasis. If urine pH exceeds 7.0, citrate doses must be reduced. Patients with distal tubular acidosis are an exception as their urine pH value is already high before treatment initiation.

Patients intolerant of potassium citrate may ingest citrus fruit. Wabner and Pak showed that drinking orange juice increases the urine citrate concentration to a similar degree to potassium citrate but, by contrast with the drug, does not reduce calcium oxalate excretion⁽⁵⁸⁾. Grapefruit juice, on the other hand, does not reduce risk factors for urolithiasis⁽⁵⁹⁾, and may exert adverse effects on drug metabolism by acting on cytochrome P450. Data on the effects of drinking lemonade (water with lemon juice) seem unclear^(60,61). Some researchers note its positive effects on citrate excretion, while others show no such benefits. Lemon juice contains citric acid, which is different from orange and grapefruit juices that contain high amounts of potassium citrate. A large amount of proton ion exerts a negative effect on urine properties and neutralises the positive effect of citrates.

Conflict of interest

The authors do not report any financial or personal affiliations to persons or organisations that could adversely affect the content of or claim to have rights to this publication.

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